

Ex 6.8(a) (7th) Ex. 4.5(a) (8th) CLARIFICATION

you are given

$$\log(p/\text{Torr}) = 7.960 - \frac{1780}{(T/\text{K})}$$

→ note that the variables are divided by the units associated with them, because the log must be dimensionless (i.e., no units)
so, where does this equation come from?

$$\frac{dp}{dT} = \frac{\Delta_{\text{vap}}S}{\Delta_{\text{vap}}V} = \frac{\Delta_{\text{vap}}H}{T\Delta_{\text{vap}}V}, \quad \text{and, since } V_m(g) \gg V_m(l) \\ \Delta_{\text{vap}}V \approx V_m(g)$$

$$\frac{dp}{dT} = \frac{\Delta_{\text{vap}}H}{TV_m(g)} = \frac{\Delta_{\text{vap}}H}{T(RT/p)} \quad \text{now, isolate } p\text{'s at } T\text{'s on opposite sides of the eqn'}$$

$$\frac{dp}{p} = \frac{\Delta_{\text{vap}}H}{RT^2} dT \quad \text{note that } d \ln p = \frac{dp}{p}$$

$$d \ln p = \frac{\Delta_{\text{vap}}H}{RT^2} dT \quad \text{now, you are set to integrate, but do so without limits}$$

$$\ln p = \int \frac{\Delta_{\text{vap}}H}{RT^2} dT = \frac{\Delta_{\text{vap}}H}{R} \int \frac{1}{T^2} dT$$

$$\ln p = -\frac{\Delta_{\text{vap}}H}{RT} + \text{constant}$$

- now this equation resembles the one you were given, except it is \ln instead of \log . Recall that

$$\frac{\ln x}{\log x} = 2.302585... = 2.303$$

(i) so, $\ln x = 2.303 \log x$

thus: $\ln(p/\text{Torr}) = 1833 - \frac{4099}{(T/\text{K})}$

$$\text{and } d \ln p = \frac{4098}{T^2} = \frac{\Delta_{\text{vap}} H}{RT^2}$$

$$\Delta_{\text{vap}} H = (4098)(8.314 \text{ J K}^{-1} \text{ mol}^{-1}) = \boxed{34.07 \text{ KJ mol}^{-1}}$$

$$\text{(ii) Find } T_b: \quad \log(760) = 7.960 - \frac{1780}{(T/K)} \quad T = T_b$$

$$(T_b/K) = \frac{-1780}{\log(760) - 7.960} = 350.4 \quad \text{so } \boxed{T_b = 350.4 \text{ K}}$$